

EE143 – Fall 2016 Microfabrication Technologies

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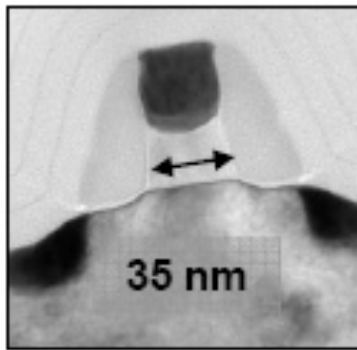
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Evolution of Devices



Yesterday's Transistor (1947)



Today's Transistor (2006)

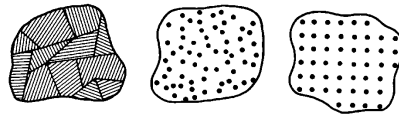


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Why “Semiconductors”?

- Conductors – e.g. Metals
- Insulators – e.g. Sand (SiO_2)
- Semiconductors
 - Conductivity between conductors and insulators
 - Generally crystalline in structure
 - In recent years, non-crystalline semiconductors have become commercially very important



Polycrystalline amorphous crystalline



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What are semiconductors?

Roman numeral gives valence

Periodic Table of the Elements

1	2											3	4						
1	H	IA											2	He					
2	3	4											5	6	7	8	9	10	
	Li	Be											B	C	N	O	F	Ne	
3	11	12	IIIA	IVB	VB	VIB	VII	VIII	IX	X	IB	IIIB	IVB	V	VI	VII	VIII	IX	X
	Na	Mg	Al	Si	P	S	Cl	Ar											
4	19	20	21	22	23	24	25	26	27	28	29	30	31	32	33	34	35	36	
	K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr	
5	37	38	39	40	41	42	43	44	45	46	47	48	49	50	51	52	53	54	
	Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe	
6	55	56	57	72	73	74	75	76	77	78	79	80	81	82	83	84	85	86	
	Cs	Ba	*La	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn	
7	87	88	89	104	105	106	107	108	109	110	111	112	113						
	Fr	Ra	+Ac	Rf	Ha	Sg	Ns	Hs	Mt	110	111	112	113						

Elements: Si, Ge, C
 Binary: GaAs, InSb, SiC, CdSe, etc.
 Ternary+: AlGaAs, InGaAs, etc.

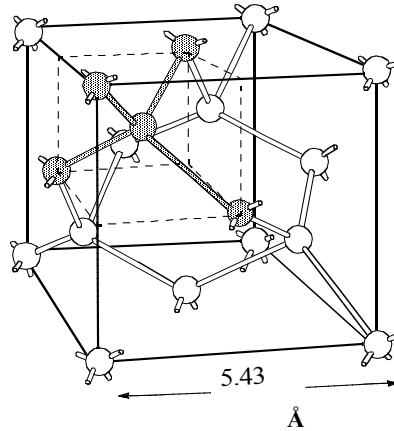


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Silicon Crystal Structure

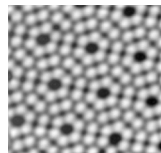
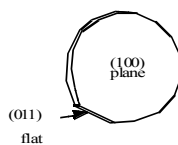
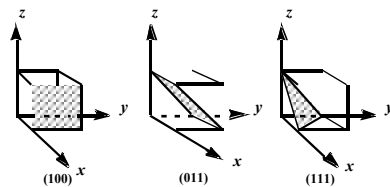
- Unit cell of silicon crystal is cubic.
- Each Si atom has 4 nearest neighbors.



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Silicon Wafers and Crystal Planes



Si (111) plane

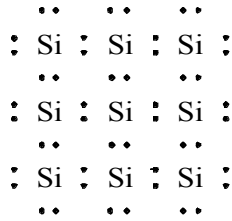
- The standard notation for crystal planes is based on the cubic unit cell.
- Silicon wafers are usually cut along the (100) plane with a flat or notch to help orient the wafer during IC fabrication.



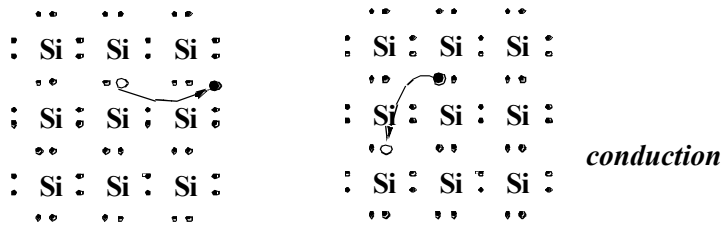
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Bond Model of Electrons and Holes (Intrinsic Si)



Simplified view of silicon crystal in a two-dimensional representation.



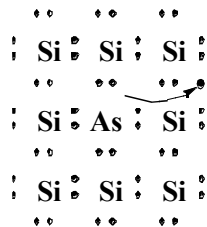
When an electron breaks loose and becomes a conduction electron, a "hole" is also created



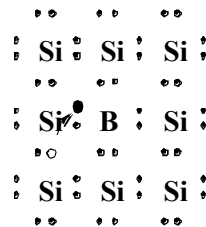
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Dopants in Silicon



N-type Si



P-type Si

- As (Arsenic), a Group V element, introduces conduction electrons and creates N-type silicon, and is called a donor.
- B (Boron), a Group III element, introduces holes and creates P-type silicon, and is called an acceptor.
- Donors and acceptors are known as dopants.



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Types of charges in semiconductors



Hole



Electron



Ionized
Donor



Ionized
Acceptor

Mobile Charge Carriers
they contribute to current flow
with electric field is applied.

Immobile Charges
they **DO NOT**
contribute to current flow
with electric field is applied.
However, they affect the
local electric field



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Doped Si and Charge

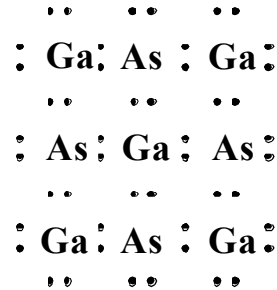
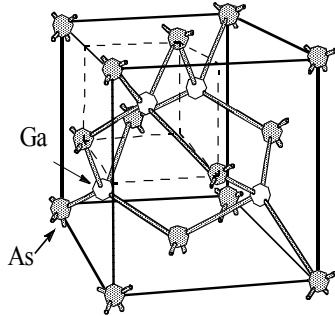
- What is the net charge of your Si when it is electron and hole doped?



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GaAs, III-V Compound Semiconductors, and Their Dopants



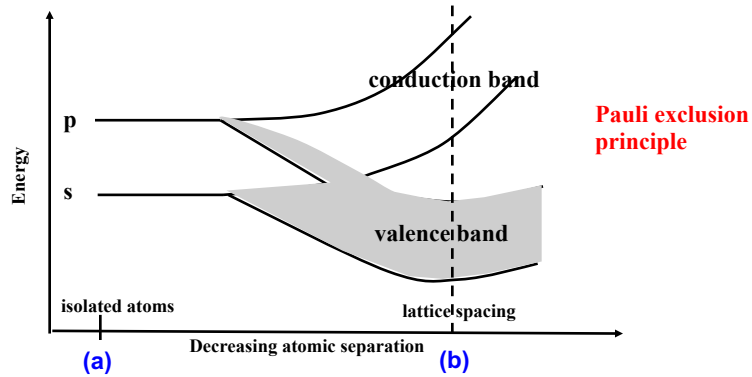
- GaAs has the same crystal structure as Si.
- GaAs, GaP, GaN are III-V compound semiconductors, important for optoelectronics.
- Which group of elements are candidates for donors? acceptors?



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From Atoms to Crystals



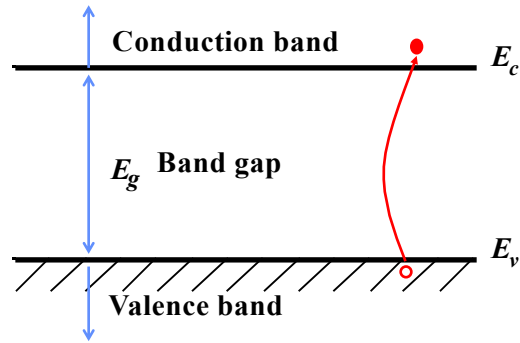
- Energy states of Si atom (a) expand into energy bands of Si crystal (b).
- The lower bands are filled and higher bands are empty in a semiconductor.
- The highest filled band is the *valence band*.
- The lowest empty band is the *conduction band*



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Energy Band Diagram



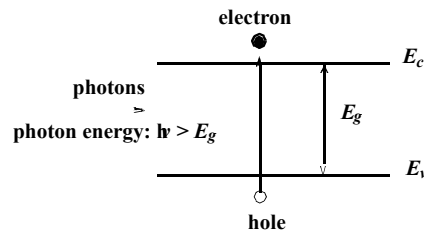
- Energy band diagram shows the bottom edge of conduction band, E_c , and top edge of valence band, E_v .
- E_c and E_v are separated by the band gap energy, E_g .



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Measuring the Band Gap Energy by Light Absorption



- E_g can be determined from the minimum energy ($h\nu$) of photons that are absorbed by the semiconductor.

Bandgap energies of selected semiconductors

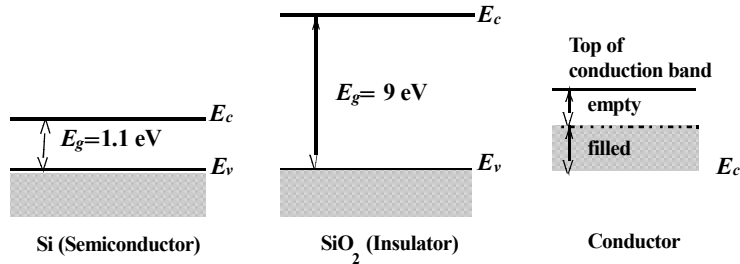
Material	PbTe	Ge	Si	GaAs	GaP	Diamond
E_g (eV)	0.31	0.67	1.12	1.42	2.25	6.0



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Semiconductors, Insulators, and Conductors



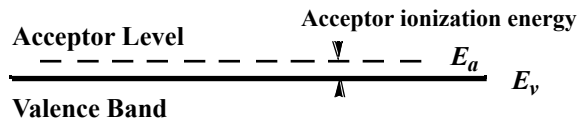
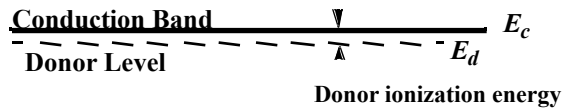
- **Totally filled bands and totally empty bands do not allow current flow.**
 - **Just as there is no motion of liquid in a totally filled or totally empty bottle.**
- **Metal conduction band is half-filled.**
- **Semiconductors have lower E_g 's than insulators and can be doped**



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Donor and Acceptor Levels in the Band Model



Ionization energy of selected donors and acceptors in silicon

Dopant	Donors			Acceptors		
	Sb	P	As	B	Al	In
Ionization energy, $E_c - E_d$ or $E_a - E_v$ (meV)	39	44	54	45	57	160

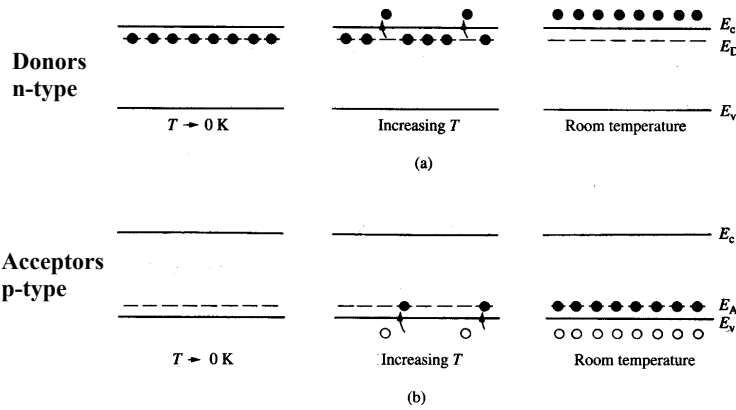
$$\text{Hydrogen: } E_{ion} = \frac{m_0 q^4}{8\epsilon_0^2 h^2} = 13.6 \text{ eV}$$



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Dopants and Free Carriers



Dopant ionization
energy $\sim 50\text{meV}$ (very low).



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General Effects of Doping on n and p

Charge neutrality: $n + N_a^- - p - N_d^+ = 0$

N_a^- : number of ionized acceptors /cm³

N_d^+ : number of ionized donors /cm³

Assuming total ionization of acceptors and donors:

$n + N_a - p - N_d = 0$

N_a : number of acceptors /cm³

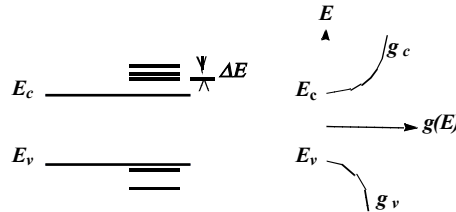
N_d : number of donors /cm³



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Density of States



$$g_c(E) = \frac{\text{number of states in } E}{E \text{ volume}} = \frac{1}{\text{eV cm}^3}$$

$$g_c(E) = \frac{m_n^* \sqrt{2m_n^* (E - E_c)}}{2\hbar^3}$$

$$g_v(E) = \frac{m_p^* \sqrt{2m_p^* (E_v - E)}}{2\hbar^3}$$



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Thermal Equilibrium

-
- No external forces applied:
 - electric field = 0
 - magnetic field = 0
 - mechanical stress = 0
 - Thermal agitation → electrons and holes exchange energy with the crystal lattice and each other
 - ⇒ Every energy state in the conduction and valence bands has a certain probability of being occupied by an electron

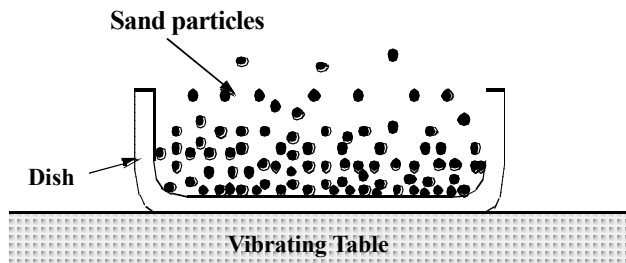


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Thermal Equilibrium

An Analogy for Thermal Equilibrium



There is a certain probability for the electrons in the conduction band to occupy high-energy states under the agitation of thermal energy (vibrating atoms, etc.)



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Fermi-Dirac Distribution

Fermi Function

Probability that an available state at energy E is occupied:

$$f(E) = \frac{1}{1 + e^{(E-E_F)/kT}}$$

E_F is called the *Fermi energy* or the *Fermi level*

There is only one Fermi level in a system at equilibrium.

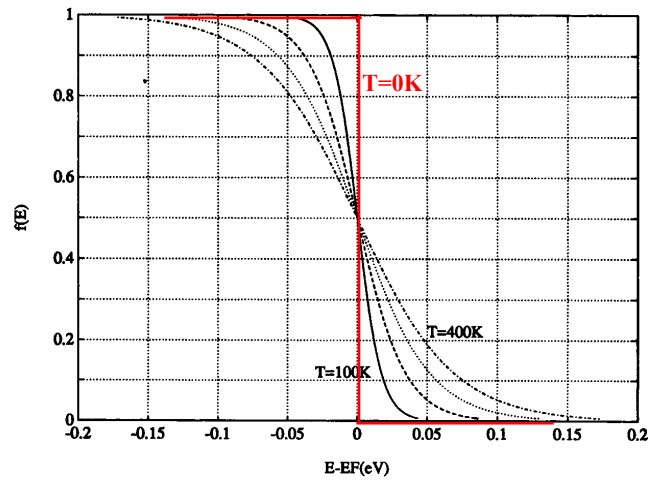
$$\text{At } E=E_F, f(E)=1/2$$



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Effect of Temperature on $f(E)$



Cal

1-23

BSAC

Question

- If $f(E)$ is the probability of a state being occupied by an electron, what is the probability of a state being occupied by a hole?

Cal

1-24

BSAC

Carrier Concentration at Equilibrium: Electrons

Equilibrium Carrier Concentrations

- Integrate $n(E)$ over all the energies in the conduction band to obtain n

$$n = \int_{E_c}^{\text{top of conduction band}} g_c(E) f(E) dE$$

- By using the Boltzmann approximation, and extending the integration limit to ∞ , we obtain

$$n = N_c e^{-(E_c - E_F)/kT} \quad \text{where} \quad N_c = 2 \left(\frac{2\pi m_n^* kT}{h^2} \right)^{3/2}$$

N_c is called the *effective density of states (of the conduction band)*.



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Carrier Concentration at Equilibrium: Holes

- Integrate $p(E)$ over all the energies in the valence band to obtain p

$$p = \int_{\text{bottom of valence band}}^{E_v} g_v(E) [1 - f(E)] dE$$

- By using the Boltzmann approximation, and extending the integration limit to $-\infty$, we obtain

$$p = N_v e^{-(E_F - E_v)/kT} \quad \text{where} \quad N_v = 2 \left(\frac{2\pi m_p^* kT}{h^2} \right)^{3/2}$$

N_v is called the *effective density of states of the valence band*.



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Intrinsic Semiconductor

- Extremely pure semiconductor sample containing an insignificant amount of impurity atoms.

$$n = p = n_i$$

E_f lies in the middle of the band gap

Material	Ge	Si	GaAs
E_g (eV)	0.67	1.12	1.42
n_i (1/cm ³)	2×10^{13}	1×10^{10}	2×10^6

